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out of the crystal pdf

2. Crystal Structure crystalline solid " the atoms or ions arrange in a pattern that repeats itself in three dimensions to form a solid which has long-range order amorphous solid " materials with only short-range order space lattice " a network composed of an infinite three-dimensional array of points

2. Crystal Structure - Rutgers Physics & Astronomy

The periodicity of atoms in crystalline solids can be described by a network of points in space called lattice. A space lattice can be defined as a three dimensional array of points, each of which has identical surroundings. $u = ua$. $uv = ua + vb$ will repeat the point along a 2D plane, where u and v are integers.

Crystal Structure - NPTEL

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Crystal structure: the manner in which atoms, ions, or molecules are spatially arranged. Unit cell: small repeating entity of the atomic structure. The basic building block of the crystal structure. It defines the entire crystal structure with the atom positions within.

Academic Resource Center - Illinois Institute of Technology

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The crystal structure will not distort without distortion if and only if the lattice period divides $\hat{t}^2 = 1/T$ evenly. Thus a signature for time crystalline states is some approximately periodic behavior in $1/T$, especially for small T .

Time Crystals - Centre for Theoretical Cosmology Home page

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